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OPTIMIZING CHLORINE DISINFECTION BY CHLORINE INJECTION LOCATION AND PIPE DIAMETER SELECTION IN A WATER DISTRIBUTION SYSTEM

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OPTIMIZING CHLORINE DISINFECTION BY CHLORINE INJECTION LOCATION AND PIPE DIAMETER SELECTION IN A WATER DISTRIBUTION SYSTEM

By Margaret "Rita" Neff

A THESIS

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Department of Civil and Environmental Engineering

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List of abbreviations

- Cl Chlorine (mg/L)
- Ct Concentration multiplied by time (min*mg/L)
- EPA Environmental Protection Agency
- MINSA Ministerio de Salud del Peru The Peruvian Ministry of Health
- NGO Non-Governmental Organization
- WHO World Health Organization

Abstract

Due to a variety of factors chlorination is common water disinfectant for communityscale distribution systems. The town of Suyo, Peru was already equipped with a gravity fed water system and drip chlorinator which achieved 4-log inactivation of viruses throughout the town, however, it was not providing 3-log inactivation of giardia cysts at all points of the system.

This project used an EPANET model to determine and compare the potential benefits of changes which could be made to the pre-existing water system in Suyo.

The model was used to compare current operating conditions of the Suyo system, 1.5 mg/L chlorine addition at the reservoir, with different potential operating configurations. To determine if there was a more efficient concentration of chlorine addition or point of chlorination, chlorination of different concentrations taking place at the reservoir and six other points between the reservoir and the town were modeled as well as the potential to chlorinate at two points simultaneously. To increase the amount of time the chlorine had to react with the water, the pipe between the reservoir and the town was modeled at different diameters.

Replacing the pipe between the reservoir and the town from the current 4" to 8" was found to raise Ct values and achieve 3-log giardia inactivation throughout the town. No benefit was found in moving the point of chlorination closer to the city nor in chlorinating both at the reservoir and at an additional node. Increasing the concentration of chlorination was found to provide 3-log giardia inactivation to a larger percentage of the town but did not provide full coverage for all users.

1 Introduction

Since the acceptance of the germ theory of disease, clean drinking water has been widely recognized as one of the most fundamental and effective strategies to ensure and improve the health of a population. In the past few decades tremendous strides have been taken to improve the situation worldwide. In 2005 95.8% of the world's urban and 75.7% of the rural population were using improved drinking water, water from a source protected from contamination. By 2017 this had increased to 96.4% in urban areas and 84.5% in rural (UNSD 2018). As the most basic water demands are met more and more countries, communities, and individuals are looking for ways to improve preexisting water systems.

The primary goals of a water treatment system are 1) that it should improve the quality and or accessibility of water and 2) that the system must be sustainable, able to be operated and maintained by the community after construction without dependence on outside sources of funding. Sustainability is an often overlooked concern for projects in the developing world where it is arguably even more important as financial aid from governments and non-governmental organizations (NGOs) can be less dependable due to increased political instability (Johnson et al. 2008). If these outside sources are cut off when the community is unable to sustain the system themselves, then there are good odds the project will break down, resulting in nothing more useful than a waste of resources. For example, a town in Peru had most of a brand new water treatment plant built to utilize multiple basin filtration which was abandoned before completion when changes in the political landscape caused funding to be diverted to other projects. Less obvious problems include failing to provide adequate training to operators and lack of community investment (Smith 2011).

To reduce these risks and provide value to preexisting systems, project planners should consider the possibilities of a modular approach by which a system is planned and then constructed one part at a time with each completed stage able to function and improve the community even if subsequent stages fall through.

This paper focuses on one such strategy, the efficacy of alternatives to improving chlorination disinfection in a preexisting piped water system. However, as different strategies may be more effective in different situations this introductory section will begin with a review of other potential treatment options.

1.1 Water Treatment in the Developing World

Water treatment systems can be broadly classified by the location where treatment occurs. Point-of-use treatment occurs in the location, typically the home, where the water is to be used. Point-of-use treatments have the benefits that they reduce the likelihood that the water will be recontaminated between treatment and consumption. They can be very inexpensive and may be the only option for populations in remote areas. The main weakness is the strong educational component required. All people who will be treating

water must be properly educated to ensure that no well-meaning mistakes are made. Solar disinfection and boiling are methods which work better for point-of-use treatment than in a larger scale system.

Solar disinfection (SODIS) is accomplished by placing clear water in a clear container, usually a bottle of polyethylene terephthalate (PET) plastic, and placing it in the sun for 6 hours or multiple days if there is cloud cover. The disinfection is two-fold: a combination of UV irradiation and the disinfection achieved by heating the water. For the heating component to be effective, the water temperature must be raised above the optimum microbial growth temperature. The process can also be hampered by turbidity which partially blocks irradiation and decreases the efficiency of disinfection (Vivar et al. 2017). This is a time consuming process and can be difficult to teach. One large scale version using photovoltaics was constructed to simultaneously treat water and generate electricity (Jin et al. 2016).

Boiling is the process of heating the water beyond what the harmful microbes can survive. This is very effective and widely practiced in Peru as a point-of-use treatment (Clasen et al. 2008).

Large scale systems are those which provide water to communities larger than a family group. Treatment takes place at a central location from which the water is transferred to individual consumers. This transfer can be provided by a piped system leading to individual houses or communal taps or water can be carried in trucks, jugs, or other containers. If the water is not piped directly to the consumer the potential vectors for contamination increase, particularly where containers used to transfer it are unclean or uncovered (Wright et al. 2004). The reliability of the water system is also greatly important. One study of two Uganda water systems showed that if supply was disrupted, incidence of cryptosporidium and rotavirus would skyrocket. The same trend was observed in E. coli but to a lesser degree (Hunter et al. 2009).

When there is no contamination between the treatment and the consumers, these systems provide a much greater ease of use for the consumers. Large scale systems also have the advantage that they are easier to maintain and inspect to ensure correct function.

Filtration and chlorination can be effective treatment strategies in both point-of-use and large scale situations. Filtration involves passing water through a semi-porous media which destroys or inactivates pathogens usually through the development of a biological layer but more modern filters may make use of colloidal silver impregnated ceramics. To increase sustainability, locally available filtration media are preferred. Perhaps the most basic form of filtration is to use the natural properties of the earth by building a well or pumping water out from the ground (Sharma and Amy 2009, Ahmed et. al. 2017). Traditional slow sand filters rely on sand and pebbles. Biochar, created through burning locally available organics such as farm wastes, has also been shown to be effective, dependent on their pore size and capacity, for the removal of inorganics and heavy metals, anionic contaminants, radionuclides, and arsenic (Gwenzi et al. 2017). Nonsynthetic filters are associated with low material costs and high labor requirements while still being attainable for a community.

1.2 Chlorination in the Developing World

Solid form chlorine (granular or tablet) is widely popular in the developing world due to its availability, effectiveness, cost, and relative ease of transportation and use. Chlorine interacts with water to produce hypochlorous acid, HOCl. A second reaction then forms between H⁺ and hypochlorite ion OCl⁻. HOCl and OCl⁻ are known as 'free chlorine'. The concentration of free chlorine remaining in the water after initial disinfection is the chlorine residual (Ratnayaka et al. 2017). Residual chlorine continues to provide disinfection in the water system, reducing the risks of recontamination between treatment and consumption.

> $Cl² + H₂O = HOCl + HCl$ $HOCI = H^+ + OCI^-$

Other substances present in the water can also interact with chlorine, reducing the amount of free chlorine produced thereby reducing disinfection efficiency. Ammonia will react with chlorine to form monochloramine $NH₂Cl$, dichloramine $NHCl₂$, and trichloramine NCl₃. NH₂Cl and NHCl₂ are known as 'combined chlorine' and are significantly less effective disinfectants than free chlorine. Organic compounds, common in surface waters, will interact to form disinfection byproducts which are hazardous to human health such as chloroform and haloacetic acids (Ratnayaka et al. 2017).

1.2.1 Drip Chlorination

Drip chlorinators consist of a tank holding a chlorine solution, most frequently created by mixing granular chlorine with water, which drips down into a water reservoir before the contained water flows into the water system. Depending on the residence time, a drip chlorinator may improve disinfection by allowing the chlorine more time to react with the water.

1.2.2 In-Line Chlorination

In-line chlorinators add chlorine to water as it moves through the pipe. One method of accomplishing this is to add a chamber containing a chlorine tablet to the pipe. As the water flows over the tablet it erodes, releasing chlorine into the water. These systems are inflexible. They are constructed to meet expected flow rates. If flow rates change later there is no way to adjust it afterwards.

Even when the flow rate is steady, which is by no means guaranteed, the erosion of the tablet is irregular (Yoakum, 2013). Research has demonstrated that much of this

unreliability can be improved by proper storage of the tablets. If kept in individual watertight wrapping until use, relatively consistent chlorination with free chlorine concentrations of less than 0.2 mg/L could be achieved. If exposed to humidity beforehand however, they dissolved much more quickly resulting in high concentrations during the first day and almost nothing on subsequent days. The residual chlorine concentration could be increased with the use of multiple tablets, but the effects only lasted for the first day (Orner 2011).

Similar results were found by a point-of-use study of a combination of filtration through a colloidal-silver-impregnated ceramic pot filter followed by chlorination through dissolution of calcium hypochlorite tablets. Chlorine residuals started above 2.5 mg/L on the first day and dwindled to around 0.7 mg/L over the course of the next seven days (Cash-Fitzpatrick et al. 2008).

Alternatively, chlorine powder could be mixed into a liquid solution and attached to the pipe on the downstream side of an orifice plate. The pressure differential between the two sides of the plate creates a vacuum which then draws the liquid into the pipe. Such a system can be surprisingly inexpensive to construct (Schuhmann and Karlheim 2012). The resultant hydrodynamic cavitation has been found to increase disinfection rates and reduce production of chloroform when used in conjunction with a sodium hypochlorite solution (Wang et al. 2015). It is unknown whether there has been any study into the long-term durability of these systems.

1.2.3 Diffusion Chlorination

Diffusion chlorinators function similarly to the in-line tablet chlorinators in that they consist of solid chlorine disbursed into water through erosion. Chlorine tablets are placed in a porous container which is then suspended below the water level in a reservoir. As with drip chlorination this can provide longer disinfection times.

1.3 Water Treatment in Peru

In January of 1991, cholera swept through South America. By the end of December 321,334 cases and 2,906 deaths had been reported in Peru alone (Ries et al. 1992) with a total of 533,000 cases and 4,700 deaths across the 19 countries affected (Swerdlow et al. 1992). Analyses of the epidemic showed a strong correlation between infection and drinking unboiled water, drinking water stored in contaminated containers, and eating or drinking items prepared in uncontrolled conditions such as street stalls. It has even been theorized that the warmer conditions of an El Nino year may have contributed by making the environment more hospitable for the cholera bacteria (Tickner and Gouveia-Vigeant 2005, Fraser 2009). This theory is supported by a study showing a 200% increase in hospital admission for diarrheal illness in children under 10 during the 1997 – 1998 El Nino (Checkley et al. 2017). It is also frequently attributed to lack of chlorination and poor sanitary conditions.

At the time, a study of some of the municipal wells and private water taps in the city of Piura found that a majority of those sampled had no residual chlorine (Reis et al. 1992). The city of Trujillo had no chlorination at all (Swerdlow et al. 1992).

The water treatment situation in Peru has greatly improved sine 1991 although much improvement is still needed. In 2015, 95% of the Peruvian urban population and 72% of the rural population were using an improved drinking water source (WHO 2016).

1.3.1 Rural Water Systems

Peruvian communities of fewer than 2,000 people are required to manage and maintain their own water system through the direction of a committee of six local volunteers elected for two-year terms by the community. At least two members of this committee must be female. All members are elected at the same time which sometimes results in a complete loss of knowledge of how to manage the system. Most communities reduce this risk by paying a permanent water operator to handle repair and maintenance. The committee itself is responsible for sharing information with and getting input from the community, collecting the monthly water fee paid by each household attached to the water system, and coordinating larger-scale repairs and annual maintenance work days.

These committees are in turn overseen by the Area Technica Municipal (ATM) who works out of the district municipality. In the district of Suyo there were over 60 town and annexes, many of which were difficult to get to. With limited transportation options this can result in practically no oversight or assistance to towns further away.

Many water systems are originally constructed with funds from the Peruvian government or an NGO. Operation and maintenance costs, as well as any wages paid to the operator, are paid by the committee from the monthly water fees. When the system requires repairs beyond the committee's ability to pay they must either collect a one-time additional payment from the community members or apply to the district municipality or an NGO for aid. There is frequently a lack of training for operators and committee members alike.

1.3.2 Chlorination in Peru

Peruvian water quality limits are stipulated by MINSA (the Ministerio de Salud, ministry of health) in the Reglamento de la Calidad De Agua Para Consumo Humano (MINSA 2011). Chlorine regulation consists of two parts: 90% of all measurements of chlorine residual taken throughout the water system must be 0.5 mg/L or above, the remaining 10% cannot be lower than 0.3 mg/L. The minimum water quality regulations mandated by MINSA, the World Health Organization (WHO), and the US Environmental Protection Agency (EPA) are shown in Table 1-1 below.

	2010 I Peru MINSA	WHO	EPA
Chlorine Residual	$90\% > 0.5$	\geq 0.2 (\geq 0.5 in high	
Min (mg/L)	$10\% \ge 0.3$	risk circumstances)	
Chlorine Residual	5	5	4 (as $Cl2$)
Max (mg/L)			
Thermotolerant	θ		
Coliforms (mg/L)			
Total Coliforms	θ		0 (in 95% of
(mg/L)			monthly samples)
Turbidity (NTU)	\leq 5		\leq 5
pH	$6.5 - 8.5$		$6.5 - 8.5*$
Conductivity	< 1500		
$(\mu mho/cm)$			
Total Dissolved	≤ 1000		${}<$ 500*
Solids (mg/L)			

Table 1-1. *Minimum Water Quality Regulations (MINSA 2011) (WHO 2017) (EPA 2018)*

*Secondary standard

Like the WHO, Peru places the maximum safe limit on chlorine residual in drinking water at 5 mg/L. The EPA uses a 4 mg/L maximum (EPA 2018). Because it is more conservative, 4 mg/L has been used for the purposes of this paper.

1.4 Description of Study Area

Peru is located on the western coast of South America, bordering the Pacific Ocean to the west, Ecuador and Colombia to the north, to the east by Brasil and Bolivia, and to the south by Chile. In 2015, approximately 31,377,000 people lived within its 494,209 square mile area (WHO 2017 and CIA 2018). In 2013 its Gross national income per capita (PPP int. \$) was 11, compared to 53 in the United States in the same year. In 2015 life expectancy at birth was 73 and 78 years for men and women, respectively, 4 years lower than in the U.S. (WHO 2017).

Figure 1-1. *A map of Peru with the district of Suyo labeled (from Google Maps).*

The study area was the City of Suyo, in the northern region of Piura. It's location in Peru is shown by the map in Figure 1-1. Suyo sits only a few degrees south of the equator with an annual average temperature of 15 \degree C which climbs to 29 \degree C in the summer months. As of the 2007 Census, 11,951 people lived in the district of Suyo. Of these, 985 lived in the capital city, also named Suyo (Suyo 2012). The city of Suyo sits directly on the old panamerican highway, a well-maintained two lane highway which allows easy access to public transportation by taxi and bus to the regional capital city, Piura, to the Southwest and into Ecuador to the North. The Ecuadorian border is 15 minutes by taxi from the city of Suyo. Some towns within the Suyo district are along the highway. Those that are not are connected by dirt roads. As a general rule, the further one moves from the highway the worse the road quality becomes and the more expensive and time consuming transportation is. Spanish is the only language spoken by most residents. Roman Catholic is the predominant faith with an Evangelical presence.

Suyo sits in the foothills of Ayabaca, where the desert of lower Piura begins to transform into mountains. Many communities have spring-fed piped water systems which flow by gravity although towns in the lower elevations may have to rely on pumps and river water. Many systems include pressure-break and purge valves although not all.

1.4.1 Water Quality of Study Area

Owing to its size, the town of Suyo is capable of maintaining a water pump and, therefore, utilizes groundwater for its drinking water. This makes the influent water significantly freer of contamination than the sources used in the rest of the district, surface water, or a protected spring.

Sampling Location	Reservoir
Date	8/26/2016
Time	7:47 AM
Turbidity (NTU)	2.06
Residual Chlorine (mg/L)	0.11
рH	8.11
Conductivity $(\mu mho/cm)$	832
TDS (mg/L)	416
Total Coliforms (CFU/100ml)	0
Thermotolerant Coliforms (CFU/100ml)	∩

Table 1-1-2. *Water Quality of the Suyo Reservoir During Chlorination*

Water quality sampling was performed for multiple locations in the Suyo district August $24th - 26th$ of 2016. Table 1.2 contains the water quality measurements of the city of Suyo reservoir after chlorination. Suyo currently employs drip chlorination. Granulated chlorine is mixed with water to form a concentrated chlorine solution stored in a tank above the reservoir with a valve and pipe system to feed the solution into the reservoir. As can be seen from the table, at the time of sampling residual chlorine levels fell below the 0.3 mg/L minimum. Even so, no coliforms were detected. The water temperature was not measured.

2 Methodology

2.1 EPANET Model

A map of the Suyo water system was provided by the Suyo Municipality and transcribed into EPANET, a public domain software created and distributed by the US EPA to model drinking water distribution systems. More information can be found at <https://www.epa.gov/water-research/epanet> (Rossman 2000) as well as the program available free to download. The original map did not include the reservoir or the full length of the pipe between the reservoir and the town. The reservoir's dimensions were known from previous measurements. The distance between the reservoir and the town was approximated from GPS coordinates of the reservoir and a known point in the town. Six pipe nodes were added in the EPANET model along the pipeline between the reservoir and the town to serve as potential points for the addition of chlorine.

Individual dwellings were not given their own nodes on the model. Instead, a node was placed at the mid-point of each block. Demand for each node was calculated based on the number of homes and businesses fed from that pipeline on the block. This simplification was made to increase clarity of the model and reduce potential for errors when transcribing the data to the model. More specific data from the EPANET model appears in Appendix C.

Figure 2-1 *EPANET schematic of the Suyo water system*

EPANET models bulk flow chlorine decay with the following first order reaction:

$$
R=K_b\mathbin{\hbox{\tt\char'42}} C
$$

 $R =$ Instantaneous rate of reaction (mg/L/hour)

 K_b = Bulk reaction rate coefficient (in this case, -1.0)

 $C =$ Reactant concentration (mg/L)

Boundary layer reactions, those taking place along the pipe wall, were modeled using a reaction coefficient of 0 (Rossman 2000).

2.2 Demand

A survey conducted in the neighboring annex of El Jardin in August of 2016 found there to be approximately 4.48 people living in each house; therefore, base demand was calculated from the assumption that 5 people lived in each occupied house. The map of the Suyo water system showed 281 occupied houses. The population was therefore estimated to be 1,405 inhabitants. This figure is believed to be a reasonable estimation of population growth since the 2007 census (Suyo 2012) at which time the Suyo population was 985.

Water consumption per person was assumed to be 90 liters per capita per day (lcd) with a factor of safety of 1.56. These numbers were those used by local engineers building and designing equivalent water systems for populations in the hotter regions of Peru where the majority of the population had flush toilets, as was the case in Suyo. These figures are supported by Twort's demand estimates for developing countries (Ratnayaka et al. 2017).

2.3 Calculation of Ct Value

Calculating the concentration of the chlorine residual times the time it has had to react in the water (Ct) proved to be a challenge as EPANET reports water age from the reservoir to each node on the system but not from one node to another. One cannot, therefore, directly calculate the influence on the Ct value from chlorine added at a node partway through the system.

First, the Ct values for all runs in which all chlorine added was added at the reservoir were calculated as follows:

$$
Ctrl = Cb * WArb * 60 min/hr
$$

 C trb = Ct value at Point b due to chlorine added at the reservoir (min*mg/L)

 $Cb = Chlorine concentration at Point b (mg/L)$

 $WArb = Water Age$ from the reservoir to Point b (hrs)

The system was then used to model different concentrations of chlorine added either at the reservoir or at one of six points between the reservoir and the start of the town (Node 1). These nodes were labeled InA, InB, etc. Their locations are shown in Figure 2-2. The node of chlorination is referred to here as Point a. The node where the chlorine residual was measured is referred to as Point b. Water age from Point a to Point b was calculated by subtracting the water age from the reservoir to Point a from the water age from the reservoir to Point b using the following formula.

 $Ctab = Cb * (WArb - WAra) * 60 min/hr$

Ctab = Ct value at Point b due to chlorine added at Point a $(min*mg/L)$

 $Cb = Residual$ chlorine concentration measured at Point b (mg/L)

 $WArb = Water Age from reservoir to Point b (hrs)$

WAra = Water Age from reservoir to Point a (hrs)

Figure 2-2 *Nodes where chlorine is added and Node 1*

This base value was used to calculate the Ct value for runs where chlorine was added at both the reservoir and at another point. To get the Ct contribution from the chlorine added part way through the system (referred to here as Point a) the chlorine residual concentration was calculated by taking the chlorine residual at the point measured (Point b) when chlorine was being added at both the reservoir and Point a and subtracting what the chlorine residual had been at Point b when chlorine was only being added at the reservoir:

$$
Crab = Crb + ((Cb - Crb) * (WArb - WAra) * 60)
$$

Ctrab = Ct value at Point b due to chlorination both the reservoir and Point a $(mg * mg/L)$

Ctrb = Ct value at Point b when chlorine was only added at the reservoir ($min*mg/L$)

 $Cb = Residual$ chlorine concentration measured at Point b (mg/L)

Crb = Residual chlorine concentration measured at Point b when chlorine was only added at the reservoir (mg/L)

 $WArb = Water Age$ from the reservoir to Point c (hrs)

WAra = Water Age from the reservoir to Point a (hrs)

For the purposes of determining the minimum Ct required, the water was assumed to have a constant pH of 8 and a temperature of 10 $^{\circ}$ C. The minimum required Ct was then determined to be the Ct required for 3-log inactivation of giardia based on the minimum residual concentration found at Node 1 after stabilization.

2.4 Determination of Minimum Required Ct

Because the water in Suyo was 8.11 pH when measured it was assumed to have a pH of 8 for the purposes of determining minimum required Ct.

Water temperature was not measured. As Suyo itself has an average ambient temperature of 15 °C and the water delivered by the Suyo water system was cool to the touch, the minimum required Ct value was based on 3-log inactivation of giardia at 8 pH and an assumed water temperature of 10°C. Ct tables can be found in Appendix B.

Ct values for a 3-log removal of Giardia and a 4-log removal of viruses were available and, therefore, were used in this study to test sufficient removal of both.

2.5 Pipe Size

After modeling a variety of chlorination concentrations at different injection points it was determined that an adequate Ct value to achieve 3-log inactivation of giardia was not being achieved. A series of models were then conducted with different diameters for pipes between the reservoir and Node 3 (shown in Figure 2-1). Node 3 was chosen because it is the first branching node in the system, to prevent pressure build up within the pipeline caused by suddenly shrinking the diameter of the pipe.

2.6 Time to Stabilization

All models were run on the assumption that chlorination would begin at midnight. This is by no means representative of field conditions and was done to ensure that all models could be objectively compared.

Because chlorination began at a time with very low demand, concentration of chlorine residual through the water system is very different in the first few hours than it is for the rest of the run time. For this reason the results section examines only those chlorine residual concentrations and Ct values which occurred after chlorine residuals within the system stabilized to more typical values. 10:00 AM was used as the stabilization time save for a few scenarios studied where the most distant node required more time.

3 Results

3.1 Baseline

As the goal was to improve the performance of the system it was important to first ascertain how the EPANET model would simulate current conditions: 1.5 mg/L chlorine addition at the reservoir. The results appear below in Tables $3.1 - 3.3$. Values which fall outside of required ranges have been highlighted.

Table 3-1. *Maximum concentration of chlorine residual after stabilization due to chlorination of 1.5 mg/L at the reservoir.*

Max Chlorine Concentration at Node after Stabilization (mg/L)									
Scho	76	68	52	43	34			15	Nodes
.36		27	34	39					

Table 3-2. *Percent of hourly residual chlorine measurements which pass Peruvian regulations after stabilization of 1.5 mg/L chlorine addition at the reservoir.*

	≥ 0.5	≥ 0.3
	$m\mathbf{g}/L$	$m\Omega/L$
% Residual Measurements after Stabilization 100%		1100%

As can be seen by these results, current chlorination in Suyo presents no health risks due to over-chlorination and meets Peruvian legal requirements for minimum chlorine residual. However, Ct values are insufficient to achieve 3-log giardia inactivation although they are more than sufficient for 4-log virus inactivation, Ct tables are shown in Appendix B.

3.2 Chlorination at Different Locations

The system was then modeled with chlorine addition taking place at different nodes between the reservoir and the city. As can be seen in Figure 3.1 below, the maximum

chlorine residual measured in the water system was only slightly affected by the location where chlorination occurred.

Figure 3-1. *Maximum chlorine residual in the Suyo water system after chlorination at different nodes.*

In all cases the maximum chlorine concentration was found at Node 1, the node closest to the points of chlorination. It was found to be nearly identical to the chlorination concentration. 3.5mg/L was deemed close to the maximum acceptable value of 4mg/L without exceeding it.

		% Residual Measurements above 0.5 mg/L during									
Chlorination		Chlorination at Node									
(mg/L)	Reservoir	InF InC InE InB InD InA									
0.5	0%	0%	0%	0%	0%	0%	6%				
1	99%	100%	100%	100%	100%	100%	100%				
1.5	100%	100%	100%	100%	100%	100%	100%				
$\overline{2}$	100%	100%	100%	100%	100%	100%	100%				
2.5	100%	100%	100%	100%	100%	100%	100%				
3	100%	100%	100%	100%	100%	100%	100%				

Table 3-4. *Percent of chlorine residual measurements throughout the Suyo water system at least 0.5 mg/L after stabilization when chlorination took place at the specified node.*

Location of chlorination also had very little effect on percent measurements of chlorine residual which met the Peruvian national standards. As can be seen in Tables 3.4 and 3.5 below, for all conditions modeled chlorination of 1.0 mg/L achieved the required minimum free chlorine concentrations at all nodes after stabilization.

		% Residual Measurements above 0.3 mg/L during										
Chlorination	Chlorination at Node											
(mg/L)	Reservoir	InC InD InE InF lnB InA										
0.5	99%	99%	99%	99%	99%	99%	99%					
1	100%	100%	100%	100%	100%	100%	100%					
1.5	100%	100%	100%	100%	100%	100%	100%					
2	100%	100%	100%	100%	100%	100%	100%					
2.5	100%	100% 100% 100% 100% 100% 100%										
3	100%	100%	100%	100%	100%	100%	100%					

Table 3-5. *Percent of chlorine residual measurements throughout the Suyo water system of at least 0.3 mg/L after stabilization when chlorination took place at the specified node.*

As was the case in the baseline model, attaining a minimum Ct value for 3-log giardia removal was found to be the most limiting condition. Tables 3-6 through 3-12 show the minimum Ct value achieved after stabilization at the nodes studied under different chlorination conditions. The required minimum Ct is included in the table. Nodes which do not achieve it have been highlighted.

Table 3-6. *Minimum Ct at node specified after stabilization from different concentrations of chlorination added at the reservoir.*

							Minimum Ct Value at Node after Stabilization (min*mg/L)				
	Chlorination ∍ $\overline{\text{[mg]}}$	Scho	N76	N68	N52	N43	N34	N ₁	N2	N15	/Bu Required ರ $min*$ Min
	0.5	65	113	59	70	51	61	26	29	47	149
	1	130	225	116	140	102	124	52	57	93	158
ಕ	1.5	196	341	175	207	153	185	79	86	138	168
	$\mathbf{2}$	261	456	234	279	203	248	105	114	186	174
rvoir	2.5	325	566	293	349	255	309	131	142	232	182
Chlorination Reser	3	390	682	350	420	305	372	158	171	278	190
	3.5	456	794	409	490	356	433	184	200	326	197

	Chlorination (mg/L)		Minimum Ct Value at Node after Stabilization (min*mg/L)								
		Scho	N76	N68	N52	N43	N34	N ₁	N2	N15	(min*mg/L) Required Min
	0.5	63	111	59	68	48	61	24	26	45	149
ಕ	1	127	224	115	136	98	123	48	52	88	158
	1.5	189	335	172	206	147	185	72	79	134	168
	2	253	448	229	272	195	245	96	105	178	174
Chlorination	2.5	316	559	284	340	244	306	120	131	222	182
≦	3	379	672	340	408	293	367	144	158	265	190

Table 3-7. *Minimum Ct at node specified after stabilization from different concentrations of chlorination added at node InA.*

Table 3-8. *Minimum Ct at node specified after stabilization from different concentrations of chlorination added at node InB.*

								Minimum Ct Value at Node after Stabilization (min*mg/L)			
	Chlorination $\frac{\text{Im}g/L}{\text{Im}g}$	Scho	N76	N68	N52	N43	N34	N1	N ₂	N15	(min*mg/L) Required Min
	0.5	61	108	54	66	47	58	21	24	42	149
ಕ	1	122	220	108	131	93	116	43	47	84	158
	1.5	183	328	162	198	139	174	65	71	126	168
	2	244	442	216	264	186	231	86	95	168	174
Chlorination	2.5	305	550	270	330	232	289	107	119	209	182
8ul	3	365	659	325	395	279	347	129	143	251	190

Table 3-9. *Minimum Ct at node specified after stabilization from different concentrations of chlorination added at node InC.*

			concentrations of childrenation added at hour film.											
			Minimum Ct Value at Node after Stabilization (min*mg/L)											
	Chlorination (mg/L)	Scho	N76	N68	N52	N43	N34	N ₁	N2	N15	(min*mg, Required ರ Min			
	0.5	53	102	48	60	39	38	15	$17\,$	35	149			
$\overline{\overline{a}}$	1	107	209	95	119	78	77	29	34	71	158			
	1.5	160	313	144	178	118	116	44	51	106	168			
	2	213	420	191	236	157	154	59	68	141	174			
Chlorination	2.5	266	524	239	295	196	193	73	85	177	182			
	3	320	628	286	355	236	231	88	102	212	190			

Table 3-10. *Minimum Ct at node specified after stabilization from different concentrations of chlorination added at node InD.*

Table 3-11. *Minimum Ct at node specified after stabilization from different concentrations of chlorination added at node InE.*

			Minimum Ct Value at Node after Stabilization (min*mg/L)											
	Chlorination	Scho	N76	N68	N52	N43	N34	N1	N2	N15	(min*mg/L, Required ರ Nin			
	0.5	47	99	43	54	29	45	$10\,$	12	31	149			
ಕ	1	95	201	84	108	57	89	20	24	60	158			
Chlorination	1.5	142	303	127	161	86	133	29	36	92	168			
	$\overline{2}$	190	400	169	216	114	177	39	48	121	174			
	2.5	237	504	212	270	143	222	49	61	152	182			
ٵ	3	284	603	254	325	172	266	59	73	183	190			

Table 3-12. *Minimum Ct at node specified after stabilization from different concentrations of chlorination added at node InF.*

The closer to the town that chlorine was injected, the lower the Ct values achieved throughout the system. Indeed, with the original pipe diameter none of the chlorination strategies modeled were able to deliver 3-log giardia inactivation to the entire town.

3.3 Chlorination at Multiple Locations

Models were run to simulate chlorination of 0.5 mg/L at the reservoir as well as secondary chlorination at one of three nodes between the reservoir and the town. Figure 3-2 shows that, as before, the node of additional chlorination made minimal difference to the maximum chlorine residual found in the system.

Figure 3-2. *Maximum chlorine residual in the Suyo water system after chlorination of 0.5 mg/L at the reservoir and additional chlorination at different nodes*

All secondary chlorination above 0.5 mg/L achieved minimum Peruvian water quality standards for residual chlorine measurements (Tables 3-14 and 3-15).

InD, InE, and InF.											
Secondary Chlorination	and Node	% Residual Measurements above 0.5 mg/L during Chlorination at Reservoir									
(mg/L)	InD	InE InF									
0.5	75%	76%	86%								
1	100%	100%	100%								
1.5	100%	100% 100%									
ኅ	100%	100%	100%								

Table 3-13. *Percent of chlorine residual measurements at least 0.5 mg/L after stabilization from 0.5 mg/L chlorination at reservoir and additional chlorination at nodes*

Table 3-14. *Percent of chlorine residual measurements at least 0.3 mg/L after stabilization from 0.5 mg/L chlorination at reservoir and additional chlorination at node InD, InE, and* InF.

	$IID, IID, unu III$.							
		% Residual Measurements above 0.3						
Secondary		mg/L during Chlorination at Reservoir						
Chlorination	and Node							
(mg/L)	InD	InF InE						
0.5	100%	100%	100%					
	100%	100%	100%					
1.5	100%	100%	100%					
	100%	100%	100%					

As with chlorination at only a single node, however, none of the permutations of twonode chlorination achieved minimum required Ct for 3-log giardia inactivation. Again, 4 log virus inactivation was achieved (Tables 3-15 through 3-17).

Table 3-15. *Minimum Ct at node specified after stabilization from 0.5 mg/L chlorination at the reservoir and different concentrations of chlorination added at node InD.*

			Minimum Ct Value at Node (after 10a) (min*mg/L)								
	ω ┙ Chlori $\frac{3}{2}$	Scho	76	68	52	43	34	1	$\mathbf{2}$	15	(min*mg/L) Required ö Min
	0.1	75	133	68	82	59	72	29	32	54	149
٠	0.5	118	219	106	130	90	113	41	46	82	158
Booster Res		171	321	154	188	129	163	55	62	117	168
\overline{a} 을 š LN,	1.5	225	428	201	248	169	214	70	79	152	174
ಕ ᇤ 0	2	278	532	250	307	208	265	85	96	187	182

Table 3-16. *Minimum Ct at node specified after stabilization from 0.5 mg/L chlorination at the reservoir and different concentrations of chlorination added at node InE.*

		$\bm{\nu} \bm{\nu}$				Minimum Ct Value at Node (after 10a) (min*mg/L)					
	ω Chlori $\frac{1}{2}$	Scho	76	68	52	43	34	1	2	15	(min*mg/L) Required Min
	0.1	74	134	67	81	58	71	28	31	53	149
+	0.5	113	214	102	124	85	107	36	41	77	158
Booster Res		160	316	145	178	117	151	46	53	108	168
\ddot{a} 뉱 Flow LN,	1.5	207	415	188	232	150	197	56	65	138	174
ಕ o	$\overline{2}$	255	517	230	285	184	241	65		169	182

Table 3-17. *Minimum Ct at node specified after stabilization from 0.5 mg/L chlorination at the reservoir and different concentrations of chlorination added at node InF.*

3.4 Chlorination with Larger Pipe

Models were run to simulate chlorination at the reservoir with larger pipe diameters between the reservoir and Node 3 to increase chlorine contact time and therefore increase the Ct value. Velocity is less in a larger pipe since $V=Q/A$ where V is velocity, Q is discharge, and A is the pipe cross-sectional area. Simulations were run for pipe diameters of 5", 6", 7", and 8". Pipe diameter was found to make a small difference to maximum chlorine residual with smaller diameter pipes delivering higher concentrations of free chlorine as showing in Figure 3-3.

Figure 3-3. *Maximum Chlorine Residual found in the Suyo Water System after Chlorination at the reservoir with different inflow pipe diameters.*

For chlorination concentrations of 1.0 mg/L and above all studied pipe diameters delivered high enough residual chlorine throughout the water system to be consistent with Peruvian regulations (Tables 3-18 and 3-19).

All modeled pipe sizes can achieve 3-log giardia inactivation throughout the water system with a chlorination concentration of 3 mg/L or lower (Tables 3-20 through 3-23). Larger pipes deliver consistently higher Ct values and can, therefore, achieve 3-log giardia inactivation at lower chlorine concentrations.

Chlorination		% Residual Measurements above 0.3 mg/L after Chlorination with Pipe Diameter						
(mg/L)	5"	6"	7"	8"				
	100%	100%	100%	100%				
1.5	100%	100%	100%	100%				
$\overline{2}$	100%	100%	100%	100%				
2.5	100%	100%	100%	100%				
3	100%	100%	100%	100%				

Table 3-19. *Percent of chlorine residual measurements at least 0.3 mg/L after stabilization from chlorination at the reservoir with different pipe sizes.*

Table 3-20. *Minimum Ct at node specified after stabilization from chlorination at the reservoir with a 5" pipe diameter.*

								Minimum Ct at Node after Stabilization (min*mg/L)				
			1	$\overline{2}$	15	43	34	Scho	52	68	76	(min*mg/ Required ರ Min
		1	73	78	111	121	143	151	155	140	244	158
	added	1.5	110	118	167	182	212	226	237	210	372	168
		2	146	156	224	241	284	302	315	279	491	174
Pipe	Reservoir Chlorine	2.5	183	195	279	302	355	377	394	349	616	179
៲៓	ä	3	220	234	335	362	426	453	472	420	741	186

Table 3-21. *Minimum Ct at node specified after stabilization from chlorination at the reservoir with a 6" pipe diameter.*

				Minimum Ct at Node after stabilization (min*mg/L)								
			1	$\mathbf{2}$	15	43	34	Scho	52	68	76	(min*mg/L) Required ರ in
	ä	1	126	131	163	169	194	199	211	192	290	162
	(mg/L) added	1.5	188	198	245	255	288	300	321	286	436	170
		2	251	264	324	340	385	398	427	382	579	179
Pipe	Reservoir Chlorine	2.5	315	330	408	424	482	498	532	478	722	190
ミ		3	377	395	489	509	578	597	639	574	869	197

Table 3-22. *Minimum Ct at node specified after stabilization from chlorination at the reservoir with a 7" pipe diameter.*

Table 3-23. *Minimum Ct at node specified after stabilization from chlorination at the reservoir with an 8" pipe diameter.*

			Minimum Ct at Node after stabilization (min*mg/L)									っ
			1	$\overline{2}$	15	43	34	Scho	52	68	76	(min*mg/ Required ರ Nin
	ੱਕ	1	161	167	200	205	223	230	248	221	313	158
	(mg/L) added	1.5	241	250	302	310	333	345	371	332	470	170
		2	321	334	399	413	445	461	497	441	626	179
Pipe	Reservoir Chlorine	2.5	401	415	502	515	558	578	619	553	786	186
ة ∞		3	482	499	601	618	668	694	742	662	942	194

4 Conclusions and Recommendations

The Suyo water system is currently delivering water which achieves Peruvian legal standards of water quality, 4-log inactivation of viruses, and is well below EPA and WHO standards of safe free chlorine concentration. It does not, however, achieve 3-log giardia inactivation. An effective method of upgrading the current system such that 3-log inactivation is achieved would be to replace the pipe between the reservoir and the town to 8" in diameter. This would increase the ability of the chlorine to interact with the water, raising Ct values to within required levels.

Increasing the concentration of chlorination was found to provide 3-log inactivation of giardia to a larger percentage of the town but did not provide full coverage for all users.

No benefit was found in moving the point of chlorination closer to the city nor in chlorinating both at the reservoir and at an additional node. Doing so made minimal impact on maximum chlorine residual concentrations while reducing the amount of time the chlorine had to disinfect the water, thereby reducing the Ct.

5 Discussion

This study identified time as the main limiting factor of effectiveness of chlorination. It was found that lower concentrations of chlorination could be equally effective or more effective than higher concentrations as long as there was more time for the chlorine disinfect the water. The only method studied was to increase the diameter of the pipe between the reservoir and the town, however, other methods may be equally productive such as increasing the size of the reservoir to increase tank residence time.

It is also important to note that although increasing the pipe diameter improved effectiveness of chlorination, the benefit of physically replacing the pipe, at least in this instance, did not outweigh the cost. At the time of the study,the Suyo system met Peruvian legal requirements for a water system. Current chlorination practices provide sufficient protection against viruses and Giardia, used as the main indicator for Ct value, had not been an issue. When the pipe needs to be replaced due to maintenance it may be beneficial to replace it with a wider diameter but to do so before hand would be an unnecessary expense to the community.

The EPANET model itself could be improved by using field data of on-site chlorine residuals to determine more accurate coefficients to model bulk and wall reactions for chlorine decay.

It must also be kept in mind that Suyo utilizes ground water. Towns which rely on surface water will need to pay more attention to organic materials which may react with chlorine, potentially creating harmful byproducts or at least reducing the effectiveness of the chlorination.

6 Bibliography

Ahmed, A. K., Abdella, & Marhaba, T. F. (2017). Review on river bank filtration as an in situ water treatment process. *Clean Technologies and Environmental Policy*, 19(2), 349- 359.

Cash-Fitzpatrick, D. (2008). *Efficacy of gravity-fed chlorination system for communityscale water disinfection in northern Ghana* (Unpublished master's thesis). Massachusetts Institute of Technology, Dept. of Civil and Environmental Engineering.

Checkley, W. (2000). Effects of El Niño and ambient temperature on hospital admissions for diarrhoeal diseases in Peruvian children. *The Lancet*, 355(9197), 442-450. doi:10.1016/s0140-6736(99)06215-7

CIA. (2018). The World Factbook: PERU. Retrieved April 17, 2018, from https://www.cia.gov/library/publications/the-world-factbook/geos/pe.html

Clasen, T. F., Thao, D. H., Biosson, S., & Shipin O. (2008). Microbiological Effectivness and Cost of Boiling to Disinfect Drinking Water in Rural Vietnam. London School of Hygiene and Tropical Medicine, Keppel St., London, WC1E 7H, U.K., and WHO Collaborating Centre for Water Supply and Waste Disposal, School of Environment, Resources and Development, Asian Institute of Technology, Pathumthani, Thailand. Environ. Sci. Technol., 42 (12), pp 4255–4260. DOI: 10.1021/es7024802. Publication Date (Web): February 5, 2008. Copyright © 2008 American Chemical Society

Clasen, T. F., Alexander, K. T., Sinclair, D., Boisson, S., Peletz, R., Chang, H. H., Majorin, F., and Cairncross, S. (2015). Interventions to improve water quality for preventing diarrhoea. The Cochrane Database of Systematic Reviews, (10), 1–201. Advance online publication.<http://doi.org/10.1002/14651858.CD004794.pub3>

EPA (Environmental Protection Agency). (2018, March 22). National Primary Drinking Water Regulations. Retrieved April 03, 2018, from https://www.epa.gov/ground-waterand-drinking-water/national-primary-drinking-water-regulations#one

Fraser B. J. (2009). Climate change impacts revealed: disease in Peru. *Scientific American Online*.186:983–990.

Gwenzi, W., Chaukura, N., Noubactep, C., & Mukome, F. N. (2017). Biochar-based water treatment systems as a potential low-cost and sustainable technology for clean water provision. *Journal of Environmental Management*, 197, 732-749. doi:10.1016/j.jenvman.2017.03.087

Hunter, P. R., Zmirou-Navier, D., & Hartemann, P. (2009). "Estimating the Impact on Health of Poor Reliability of Drinking Water Interventions in Developing Countries." *Science of the Total Environment,* 407, 2621-2624.

Jin, Y., Wang, Y., Huang, Q., Zhu, L., Cui, Y., & Cui, L. (2016). The performance and applicability study of a fixed photovoltaic-solar water disinfection system. *Energy Conversion and Management*, 123, 549-558. doi:10.1016/j.enconman.2016.06.073

Johnson, D. M., Hokanson, D. R., Zhang, Q., Czupinski, K. D, Tang, J. (2008). "Feasibility of Water Purification Technologies in Rural Areas of Developing Countries." *Journal of Environmental Management,* 88, 416-427.

MINSA (Ministerio de Salud del Peru) (2011) Reglamento de la Calidad de Agua para Consumo Humano. DS Nº031-2010-SA, Titulo IX: Requisitos de Calidad del Agua Para Consumo Humano, Articulo 66[°] Control de Desinfectante

Orner, K. (2011). "*Effectiveness of In-Line Chlorination of Gravity Flow Water Supply in Two Rural Communities in Panama.*" (Unpublished Masters Thesis) University of South Florida, Dept. of Civil and Environmental Engineering.

Ratnayaka, D. D., Brandt, M. J., Johnson, K. M., & Elphinston, A. (2017). *Tworts water supply*. Elsevier, Butterworth-Heinemann.

Ries, A. A., Vugia, D. J., Beingolea, L., Palacios, A. M., Vasquez, E., Wells, J. G., Baba, N. G., Swerdlow, D. L., Pollack, M., Bean, N. H., Seminario, L., Tauxe, R. V. (1992). "Cholera in Piura, Peru: A Modern Urban Epidemic." The Journal of Infectious Diseases, vol. 166, no. 6, 1992, pp. 1429–1433. JSTOR, JSTOR, www.jstor.org/stable/30113051.

Rossman, L.A. (2000). EPANET 2 User's Manual. Environmental Protection Agency, https://nepis.epa.gov/Adobe/PDF/P1007WWU.pdf

Schuhmann, R. J. and Karlheim, L. M. (2012). "*A Low-Cost Flow-Dependent Chlorine Injector for Use in Rural Developing World Water Systems*." Conference Paper, Environmental & Water Resources Institute (EWRI) of ASCE, 5th International Perspectives on Water Resources & the Environment Conference (IPWE), At Marrakech, Morocco

Sharma, S.K. and Amy, G. (2009). Bank Filtration: A Sustainable Water Treatment Technology for Developing Countries, 34th WEDC International Conference, Addis Ababa, Ethiopia.

Smith, G. (2011). *Rural water system sustainability : a case study of communitymanaged water systems in Saramaka communities*. (Unpublished Masters Thesis). Michigan Technological University, Dept. of Civil Engineering.

Suyo (Municipalidad Distrital de Suyo) (2012). Plan de Desarollo Concertado 2012 – 2021.

Swerdlow, D. L., Mintz, E. D., Rodriguez, M., Tejada, E., Ocampo, E., Espejo, L., Greene, K. D., Saladana, W., Seminario, L., Tauze, R. V., Wells, J. G., Bean, N. H., Ries, A. A., Polack, M., Veritz, B., and Blake, P. A. (1992). Waterborne transmission of epidemic cholera in Trujillo, Peru: Lessons for a continent at risk. *The Lancet*, 340(8810), 28-32. doi:10.1016/0140-6736(92)92432-f

Tickner, J. and Gouveia-Vigeant, T. (2005), The 1991 Cholera Epidemic in Peru: Not a Case of Precaution Gone Awry. *Risk Analysis*, 25: 495–502

United Nations Statistics Division (UNSD). (2018). World Data: Environment and Infrastructure Indicators. Retrieved April 07, 2018, from http://data.un.org/en/reg/g1.html

Vivar M., Pichel N., Fuentes M., López-Vargas A. (2017). Separating the UV and thermal components during real-time solar disinfection experiments: The effect of temperature. *Solar Energy*, Volume 146, 2017, Pages 334-341, ISSN 0038-092X, https://doi.org/10.1016/j.solener.2017.02.053.

Wang, Y., Jia, A., Wu, Y., Wu, C., & Chen, L. (2015). Disinfection of bore well water with chlorine dioxide/sodium hypochlorite and hydrodynamic cavitation*. Environmental Technology*, 36(4), 479-486

WHO (World Health Organization) (2016) GHO | Global Strategy for Women's, Children's and Adolescents' Health (2016-2030). Retrieved April 03, 2017, from <http://apps.who.int/gho/data/node.gswcah>

WHO (World Health Organization). (2017). Guidelines for Drinking-Water Quality, 4th Edition. Geneva.

WHO (World Health Organization) (2018). Drinking Water Fact Sheet. Retrieved April 03, 2018 from http://www.who.int/mediacentre/factsheets/fs391/en/

Wright, J., Gundry, S. and Conroy, R. (2004), Household drinking water in developing countries: a systematic review of microbiological contamination between source and point-of-use. *Tropical Medicine & International Health*, 9: 106–117. doi:10.1046/j.1365- 3156.2003.01160.xYoakum, B.A. (2013). *Improving Implementation of a Regional In-Line Chlorinator in Rural Panama Through Development of a Regionally Appropriate Field Guide.* (Unpublished Masters Thesis). University of South Florida.

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Figure 1.1: A map of Peru with the district of Suyo labeled. This figure comes from Google Maps and is fair use for research papers. Accessed April 2018.

B Ct Tables

The following tables are US EPA Ct values in min*mg/L for the inactivation of Giardia Cysts and viruses by free chlorine published by Clean Water Systems in 2007.

Free				pH			
Residual	≤ 6.0	6.5	7.0	7.5	8.0	8.5	≤ 9.0
(mg/L)							
≤ 0.4	73	88	104	125	149	177	209
0.6	75	90	107	128	153	183	218
0.8	78	92	110	131	158	189	226
1.0	79	94	112	134	162	195	234
1.2	80	95	114	137	168	200	240
1.4	82	98	116	140	170	206	247
1.6	83	99	119	144	174	211	253
1.8	88	101	122	147	179	215	259
2.0	87	104	124	150	182	221	265
2.2	89	105	127	153	186	225	271
2.4	90	107	129	157	190	230	276
2.6	92	110	131	160	194	234	281
2.8	93	111	134	163	197	239	287
3.0	95	113	137	166	201	243	292

Table [B-](#page-41-0)1 *Ct values for 3-log (99.9%) inactivation of Giardia cysts by free chlorine at water temperature 10.0 °C (50 °F)*

Table [B-](#page-41-0)2 *Ct values for inactivation of viruses by free chlorine*

Temperature		Log Inactivation							
$({}^\circ\mathrm{C})$		$2.0 - log$		$3.0 - log$	$4.0 - log$				
	pH 6-9	pH 10	pH 6-9	pH 10	pH 6-9	pH 10			
0.5		45		66		90			
		30		44		60			
10		22		33		45			
15				22		30			
20				16		22			
25									

C EPANET Model Data

For the EPANET model, the pipes within the city are labeled after the streets they run under. For example, the segments of pipe which run under Avenida Amazonas are labeled "AMAZ1", "AMAZ2", etcetera. The segments of pipe leading from the reservoir to the town are simply labeled "Pipe 1", "Pipe 2", and so on. Information used to model it appears in Tables C-1.

Pipe ID	Length (f _t)	Diameter (in)	Roughness	Bulk Coeff.	Wall Coeff.	Connected Nodes
Ama1	421	$\mathbf{1}$	100	-1	$\mathbf 0$	16,15
Amaz2	93	$\mathbf{1}$	100	-1	$\overline{0}$	17,16
Amaz3	92	$\overline{2}$	100	-1	$\mathbf 0$	17,18
Amaz4	110	$\overline{2}$	100	-1	$\mathbf 0$	19,18
Amaz5	71	$\overline{2}$	100	-1	$\mathbf 0$	20,19
Amaz6	66	$\overline{2}$	100	-1	$\mathbf 0$	20,21
Amaz7	146	$\overline{2}$	100	-1	$\mathbf 0$	22,21
Avelar	209	4	100	$^{\mbox{-}}1$	$\mathbf 0$	1,2
Bol1	138	4	100	-1	$\mathbf 0$	2,3
Bol10	198	$\mathbf{1}$	100	-1	$\mathbf 0$	66,67
Bol ₂	118	4	100	-1	$\mathbf 0$	3,12
Bol3	124	4	100	-1	0	12,19
Bol4	134	$\overline{2}$	100	$\text{-}1$	$\mathbf 0$	19,24
Bol ₅	75	$\overline{2}$	100	-1	$\mathbf 0$	24,26
Bol ₆	127	$\overline{2}$	100	-1	$\mathbf 0$	26,30
Bol7	108	$\overline{2}$	100	-1	$\overline{0}$	30,45
Bol ₈	48	$\overline{2}$	100	-1	0	45,56
Bol9	126	$\overline{2}$	100	-1	$\mathbf 0$	56,57
Cem1	173	$\overline{2}$	100	-1	$\mathbf 0$	4,3
Cem ₂	49	$\mathbf{1}$	100	-1	$\mathbf 0$	5,3
Cem ₃	276	$\mathbf{1}$	100	-1	$\overline{0}$	6,5
Cem4	183	$\mathbf{1}$	100	-1	$\mathbf 0$	6,7
CemTankLine	$\mathbf{1}$	$\overline{2}$	100	-1	$\mathbf 0$	4, CemTank
Escalera	468	$\overline{2}$	100	-1	$\mathbf 0$	34,school
Grau1	125	$\mathbf{1}$	100	$\textnormal{-1}$	$\mathbf 0$	48,49
Grau2	97	$\mathbf{1}$	100	-1	$\mathbf 0$	48,50
Grau3	133	$\mathbf{1}$	100	-1	$\mathbf 0$	50,51
Grau4	127	$\mathbf{1}$	100	-1	$\mathbf 0$	51,52

Table [C-](#page-42-0)1 *EPANET model network pipes*

Apart from the school, the reservoir, and the water tank in the cemetery the labels generated automatically by EPANET were kept as labels for the nodes. The data used to model them appears in Table C-2.

Node ID	Elevation (f _t)	Base Demand (GPM)	Initial Quality (hrs)
Junc 1	1414		
Junc 2	1414	1.4	
Junc 3	1414		
Junc 4	1420	0.38	

Table [C-](#page-42-0)2 *EPANET model network nodes*

There was one tank in Suyo. It was an open air tank in the cemetery. Information used for modeling it is in Table C-3.

Table [C-](#page-42-0)3 *EPANET model tank*

Cemetary Tank							
Label	CemTank						
Elevation	1420						
Initial Level (ft)							
Minimum Level (ft)							
Maximum Level (ft)	4.5						
Diameter (ft)	ąq						

Table [C-](#page-42-0)4 *EPANET model demand pattern*

